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~~Answering Multiple Choice Questions~~

~~3 kinds of mixture (solution, suspension, colloid)~~

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Multiple Choice Questions Solution Colloids

This set of Engineering Chemistry Multiple Choice Questions & Answers (MCQs) focuses on "Colloids". 1. What is the size of colloidal particles? a) 10-20 nm b) More than 20 nm c) Less than 10 nm d) 30 to 50 nm View Answer

Colloids Questions and Answers - Sanfoundry

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Read Book Multiple Choice Questions Solution Colloids And Suspensions endothermic can be spontaneous provided that _____. A)the solvent is a gas and the solute is a solid
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6. An example of a colloid which is an emulsion is ____ a) Whipped cream b) Mayonnaise c) Fog d) Gelatin. 7. An example of a solution is ____ a) Sugar and water b) Sand and water c) Milk d) Whipped Cream. 8. Oil floats on water. The most accurate reason for this is a) oil is less dense than water b) oil is immiscible (does not dissolve) in water

Mixtures, solutions, suspensions, colloid Quiz

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Solutions Multiple Choice Test. For your review in chemistry, you can use this 30 - item questions which I prepared for you.
1. A homogeneous mixture of two substances is a. a. colloid. c. solution. b. suspension.

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10 In the DLVO theory of colloids, normal thermal motion may be sufficient to overcome the energy barrier that ...

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by one, you will not be able to revert this action. Quiz+ | Quiz 6: Solutions and Colloids Quiz Solutions, Colloids and Suspensions : Classify the following mixtures as solutions, colloid, or suspensions - Q1: Large particles settle out on standing Suspension, Colloid, Solution,...

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1. What is the definition of lyophobic colloids? Give two examples. 2. What is the definition of lyophilic colloids? Give two examples. 3. Give an example for lyophobic sol. 4. Give an example for colloid macromolecule system. 5. Give an example for association colloid system. 6. Give an example for spongoid coherent system. 7.

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