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Molecular weight  $28.97 \text{ mol}^{-1}$

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**Solution 1.1.** To get started, first list or determine the volumes involved:  $\Delta d$  = volume of water dumped =  $100 \text{ cm}^3$ ,  $\Delta c$  = volume of a sip =  $5 \text{ cm}^3$ , and  $V_2$  = volume of water in the oceans =  $\frac{4}{3}\pi RD^3$ , where,  $R$  is the radius of the earth,  $D$  is the mean depth of the oceans, and  $\Delta$  is the oceans' coverage fraction. Here we've ignored the ocean volume occupied by salt and have assumed that the ...

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