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Examples, Ch. 21: 11-14, 17-19, 21, 24, 27-30, 59-60, 73 II.

Rates of Decay (21.4): Know the half-life equation for radioactive decay (first order reaction) and be able to use it to convert between amount and time, or vice versa.

Examples, Ch. 21: 34-42, 80 III. Energy Changes (21.6): Know how to calculate the energy change for a nuclear reaction

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this study guide covers chapter 13 (chemical kinetics),
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march 11 5 pages ch 1020 chapter 14 notes

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___A 1. A scientist is working in the lab, with compounds containing carbon. She is most likely working in the field of
The answer is A because organic chemistry is the study of most carbon-containing compounds. _A___ 2.

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