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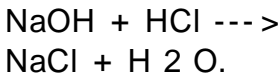
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Step 1. 1) Since
moles = molarity X
volume in L. Hence
moles of HCl mixed
= $0.35 \times 35 / 1000$
= 0.01225. And
moles of NaOH

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mixed = 0.35×45
 $/ 1000 = 0.01575$.

Since both acid and base are strong so they will react completely forming.



Since moles of NaOH > moles of HCl.

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